

Elicitation Questions

Facet Cluster 2.2: Changes in which the substance stays the same Elicitation Questions

1.	Describe the differences between the three main states of matter. (solid, liquid, and gas)
2.	Ethanol (CH ₃ CH ₂ OH), an important ingredient in bio-fuels, is a liquid at room temperature of 25°C (77°F) and turns into a gas at 78.3°C (172 °F). Carbon dioxide (CO ₂) is a gas at room temperature.
	a. Why does ethanol change into a gas at a higher temperature than CO ₂ ?
	b. What happens at the molecular level when ethanol boils?
	b. What happens at the molecular level when ethanol bons:

3. Salt dissolves in water and creates a solution that conducts electricity. Sugar dissolves in water and creates a solution that cannot conduct electricity. Provide an explanation for why this is.